Diastereoselective Addition of Organocuprates on 1,4-Dihydropyridine-3-carboxaldehydes. Synthesis of Chiral 1,4-Dihydropyridyl-3-alcohols and Pyridyl-3-alcohols

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Abstract : Chiral 1,4-dihydropyridyl-3-alcohols are obtained via a diastereosclective (de 90% -> 95%) 1,2 addition of organocuprates on chiral 1,4-dihydropyridine-3-carboxaldehydes and are transformed into chiral pyridyl-3-alcohols (90% ->95%).

We recently reported an asymmetric synthesis of 1,4-dihydropyridine-3-carboxaldehydes 1 starting from nicotinaldehyde by using chiral diamines of C_2 symmetry.¹ As a part of a program directed towards the synthetic uses of such dihydropyridines 1 in synthesis, we elected to investigate the possibility of diastereoselective transformation of the aldehyde into a secondary alcohol by addition of an organometallic reagent (scheme 1).

Initially, Grignard reagent $(CH_3MgBr)^2$ was added in ether on the dihydropyridine 1 ($R^1 = Et$, $R^3 = OCH_3$, entry 1, Table 1). The corresponding alcohol 2 was obtained in good yield, but with a poor diastereoselectivity (measured by¹H NMR). In contrast, the use of Me₂CuLi,³ in ether at - 20°C, afforded the same alcohol with an excellent diastereoselectivity (d.e. > 95%) and a good yield when the work up was performed with an aqueous solution of NH₄Cl. No product resulting from a 1,4 addition was isolated. Several dihydropyridines 1 were used, and as shown in the Table1, the diastereoselectivity was independent of the nature of R¹. With Bu₂CuLi, the same result was obtained (entry 4). In the case of Ph₂CuLi (entry 7), the reaction afforded only the racemic alcohol 3.



Entry	R ¹	R ² M	Yield(%)	d.e. ^{a)} (%)	[α] _D (CHCl ₃) ^{b)}
1	Et	MeMgBr	90	35	-
2	Et	Me ₂ CuLi	70	>95	-33 (c = 19)
3c)	Me	Me ₂ CuLi	80	90	-49 (c = 5)
4	Et	Bu ₂ CuLi	87	95	-63 (c = 8)
5	CH ₂ =CH ₂	Me ₂ CuLi	71	>95	-127 (c = 5)
6	Ph	Me ₂ CuLi	78	95	-128 (c = 3)
7	Et	Ph ₂ CuLi	-	-	-

Table 1 Addition of Organometallic Reagents on Aldehydes 1 (R³ = OCH₃)

a) Determined by ¹H NMR (400 MHz).

b) Starting from a (R) dihydropyridine 1.

c) In THF, the same d.e. was obtained in 28 % yield

d) Racemic alcohol 3 (R^1 = Ph) was isolated in 35% yield.

Dihydropyridines 2 were converted into pyridyl alcohols 3 (table 2) by alcaline treatment (scheme 2) followed by aromatization of the dihydropyridine ring.⁴ Indeed, such chiral alcohols are attractive auxiliaries used for the resolution of carboxylic acids and in the synthesis of natural products.⁵ The optical purity of the alcohols 3 was determined by 31 P NMR of the corresponding diazaphospholidines⁶ 4 (scheme 2) and found to be the same that the starting aldehyde 1.Therefore, no racemization occurs during all the procedure. The absolute configuration of the new stereogenic center was determined on alcohol 3a according Trost's procedure⁷ and found to be (S) starting from (R) dihydropyridine 1.



Scheme 2 Preparation of Pyridylalcohols 3

R ¹	R ²	alcohol	Yield(%) ^{a)}	e.e.(%) ^{b)}	(α] _D (CHCl ₃)
Me	Me	3a	68	90	-29 (c = 1)
Et ^{c)}	Me	3b	65	84	-27 (c = 1.9)
Et ^{c)}	Bu	3c	75	80	-8.4 (c = 5.2)
Vinyl	Me	3d	40	>95	-37 (c = 3.7)

a) Not optimized

Table 2

b) Determined according ref.6.

c) Starting from 1 of 86% c.e.

In order to explain the stereochemistry of the reaction, a chelation 8 of the square planar dimeric cuprate⁹ by the aldehyde and the oxygen atom of the carbamate was postulated (scheme 3). Such a chelation block the aldehyde in the conformation shown in scheme 3 and the cuprate adds on the less hindered face of the aldehyde according a Bürgi-Dunitz trajectory.¹⁰





The influence of the oxygen atom of the carbamate was demonstrated by addition of Me₂CuLi, in ether, to aldehyde 5 prepared according scheme 4.¹¹ The diastereomeric composition of the unstable alcohol 6 was measured by ¹H NMR and found to be low (34%). In contrast, the same reaction performed on aldehyde 7 afforded the corresponding alcohol with an excellent diastereoselectivity (d.e. > 95%). Therefore, the presence of a carbamate (or amide) function is necessary for the diastereocontrol.



In conclusion, we have shown that 1, 2 addition of organocuprates on 1,4-dihydropyridine-3carboxaldehydes occurs with a high diastereoselectivity. The obtained alcohols are good precursors of the corresponding 3-pyridyl alcohols. Work is in progress to exploit alcohols 2 in synthesis.

REFERENCES and NOTES.

- 1 Gosmini, R.; Mangeney, P.; Alexakis, A.; Commerçon, M.; Normant, J.F., Synlett, 1991, 111.
- 2 The use of MeLi afforded a H-N-1,4-dihydropyridine-3-carboxaldehyde resulting from a cleavage of the carbamate.
- 3 Organocopper reagents react with aldehydes at low temperature : Posner, G.H., Whitten, C.E.; Mc Farland, P.E., J.Am.Chem.Soc, 1972, 94, 5106.
- 4 An alcaline work up afforded 3-pyridyl alcohols : Comins, D.L.; Myoung, Y.C., *J.Org.Chem.*, **1990**, 55, 292.
- 5 For synthesis and use of chiral pyridyl alcohols see : Uskokovic, M.R.; Lewis, R.L.; Partridge, J.J.; Despreaux, C.W.; Pruess, D.L., J.Am.Chem.Soc., 1979, 101, 6742. Soai, K.; Hori, H.; Niwa, S., Heterocycle, 1989, 29, 2065. Seemayer, R.; Schneider, M.P., Tetrahedron Asym., 1992, 3, 827.
- 6 Alexakis, A.; Mutti, S.; Mangeney, P., J.Org.Chem., 1992, 57, 1224.
- Trost, B.M.; Belletire, J.L.; Godleski, S., Mc Dougal, P.G.; Balkovec, J.M., J.Org.Chem., 1986, 51, 2370.
- For chelated mediated addition of copper reagents to aldehydes see: a)Still, W.C.; Schneider, J.A., *Tetrahedron Lett.*, 1980, 21, 1035. b) Kunz, T.; Reissig, H.U., Angew. Chem. Int. Ed. Engl., 1988, 27, 269. c) Janowitz, A.; Kunz, T.; Handke, G.; Reissig, H.U., Synlett., 1989, 24. d) Kozikowski, A.P.; Lee, J., J.Org.Chem., 1990, 55, 863.e) Alexakis, A.; Sedrani, R.; Normant, J.F.; Mangeney, P., Tetrahedron Asym., 1990, 1, 283. f) Baldwin, S.W., McIver, J.M., Tetrahedron Lett., 1991, 32, 1937. d) Burke, S.D.; Piscopio, A.D.; Marron, B.E.; Matulenko, M.A., Pan, G., Tetrahedron Lett., 1991, 32, 857. e) Molander, G.A.; Haar, J.H., Jr, J.Am.Chem.Soc., 1993, 115, 40 and ref. cited therein.
- 9 Van Koten, G., J.Organomet.Chem., 1990, 400, 283.
- 10 Bürgi, H.B.; Dunitz, J.D.; Shefter, E., J.Am. Chem. Soc., 1973, 95, 5065.
- 11 Mangeney, P.; Gosmini, R.; Alexakis, A., Tetrahedron Lett., 1991, 32, 3981.

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